

# Thermal Analysis on Cryogenic Liquid Hydrogen Tank on an Unmanned Aerial Vehicle System

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# Thermal Analysis on Cryogenic Liquid Hydrogen Tank on an Unmanned Aerial Vehicle System

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Thermal analyses are performed on the liquid hydrogen (LH<sub>2</sub>) tank designed for an unmanned aerial vehicle (UAV) powered by solar arrays and a regenerative proton-exchange membrane (PEM) fuel cell. A 14-day cruise mission at a 65,000 ft altitude [1] is considered. Thermal analysis provides the thermal loads on the tank system and the boiling-off rates of LH<sub>2</sub>. Different approaches are being considered to minimize the boiling-off rates of the LH<sub>2</sub>. It includes an evacuated multilayer insulation (MLI) versus aerogel insulation on the LH<sub>2</sub> tank and aluminum versus stainless steel spacer rings between the inner and outer tank. The resulting boiling-off rates of LH<sub>2</sub> provided by the one-dimensional model and three-dimensional finite element analysis (FEA) on the tank system are presented and compared to validate the results of the three-dimensional FEA results. It concludes that heat flux through penetrations by conduction is as significant as that through insulation around the tank. The tank system with MLI insulation and stainless steel spacer rings results in the lowest boiling-off rate of LH<sub>2</sub>.

#### I. Introduction

The densified propellant (LH<sub>2</sub>) is stored at cryogenic conditions near its boiling point. Any heat transferring into the propellant could cause phase change such that the storage of the propellant becomes more unmanageable and unpredictable. Heat leaks into the storage tank must be minimized through proper insulation. Boiling-off rate of the propellant in the storage system has to be assessed. For a 14-day mission, minimizing the boiling-off rate of LH<sub>2</sub> could reduce the system mass significantly. Both an evacuated MLI blanket and aerogel insulation are investigated for comparison. An evacuated MLI blanket gives better thermal insulation; however, the MLI fabrication and installation requires extreme care and precision. In addition, the thermal performance of MLI is very sensitive to the vacuum pressure [2].

A representative LH<sub>2</sub> tank system is shown in Fig. 1. It consists of two concentric spherical tanks connected by spacer rings, which serve as a sleeve bearing for the support rod passing through the tank assembly. The inner tank is welded to the support rod, and the two rings are welded to both inner and outer tanks to seal the vacuum gap between the tanks where either MLI or aerogel insulation is used. The inner tank has an 8.5 ft diameter with a 0.031 in. thickness and holds 1,300 lbm of LH<sub>2</sub> stored at an internal pressure of 30 psia with 5% ullage. The outer tank is 0.25 in. thick and the MLI insulation is approximately 2 in. thick. The support rod has a 4 in. inner diameter with a 0.1 in. thickness while the ring has a 4.5 in. inner diameter with the same thickness. The support and inner/outer tanks are made of 2014—T6 aluminum and the spacer ring material is either 2014—T6 aluminum or stainless steel for comparison of heat loss.

In the following, the one-dimensional thermal circuit for the tank system is presented first, which is followed by the three-dimensional FEA on the tank system and conclusions.

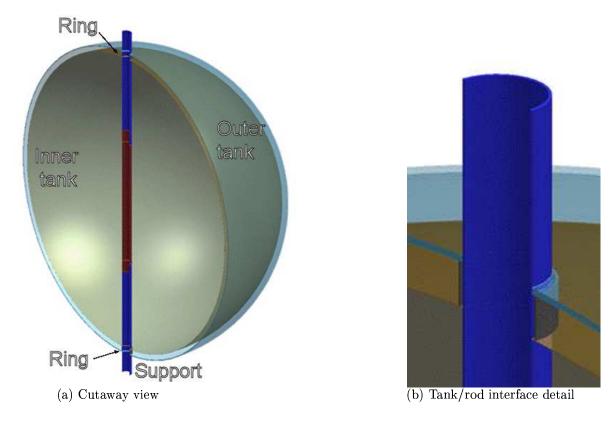


Figure 1. A representative LH<sub>2</sub> tank design (courtesy of Timothy Roach).

## II. One-dimensional Thermal Circuit for the Tank System

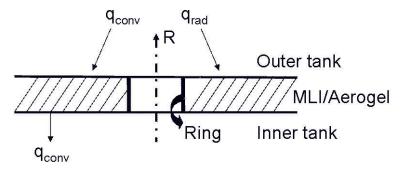
The heat transfer modes across the tank wall are sketched in Fig. 2, showing free convection and radiation outside the outer tank and free convection inside the inner tank. Note that the wall thickness of both the inner and outer tanks is ignored due to the relatively thin wall. Between the inner tank and outer tank, heat flux transfers through three parallel paths, which shows conduction through two rings and MLI/aerogel insulation. Across the evacuated MLI or aerogel insulation, both radiation and conduction take place, which is not shown in Fig. 2. Either the effective thermal conductivity,  $k_e$ , or effective heat transfer coefficient,  $h_e$ , can be used to represent the evacuated MLI or aerogel insulation [3]. The equivalent thermal circuit is plotted in Fig. 3.  $T_a$  and  $T_{LH2}$  are the temperatures of the ambient air and liquid hydrogen, respectively. The temperature of the inner and outer tanks is  $T_i$  and  $T_o$ , respectively. Assuming the radius of the inner and outer tanks being  $r_1$  and  $r_2$ , respectively, the thermal resistance between the outer tank and ambient,  $R_1$  ((W/K)<sup>-1</sup>), is defined as

$$R_1 = \left[ \frac{1}{4\pi r_2^2 h_1} + \frac{1}{4\pi r_2^2 \epsilon \sigma (T_o + T_a)(T_o^2 + T_a^2)} \right]^{-1}$$
(1)

where  $h_1$  being the heat transfer coefficient (for convection mode),  $\epsilon$  being the surface emissivity (for radiation mode), and  $\sigma$  being the Stefan-Boltzmann constant ( $\sigma = 5.67 \times 10^{-8} \text{ W/m}^2\text{-K}^4$ ); the thermal resistance across the insulation annulus (MLI/aerogel),  $R_2$  ((W/K)<sup>-1</sup>), is defined as

$$R_2 = 1/(4\pi k_e)(1/r_1 - 1/r_2) \tag{2}$$

# $T_a = 216.7 \text{ K}$ at 65,000 ft altitude



$$T_{LH2} = 20.4 \text{ K}$$

Figure 2. Schematic view of the one-dimensional heat transfer across the tank wall.

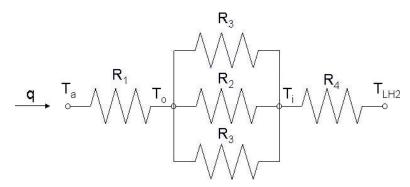


Figure 3. Equivalent thermal circuit across the tank wall.

the thermal resistance through the conduction (outer tank, ring, inner tank, and support),  $R_3$  ((W/K)<sup>-1</sup>), is defined as

$$R_3 = \frac{1}{k_1 A_{c,1} / L_1} + \frac{1}{k_2 A_{c,2} / L_2} + \frac{1}{k_3 A_{c,3} / L_3} + \frac{1}{k_4 A_{c,4} / L_4}$$
(3)

where  $k_n, L_n, A_{c,n}$  with n = 1, 2, 3, and 4 are the thermal conductivity, the length of conduction, and the contact surface area for the outer tank, the ring, the inner tank, and the support rod, respectively. The thermal resistance between hydrogen and the inner tank,  $R_4$  ((W/K)<sup>-1</sup>), is defined as

$$R_4 = 1/(4\pi r_1^2 h_4) \tag{4}$$

with  $h_4$  being the heat transfer coefficient. The total thermal resistance of the circuit,  $R((W/K)^{-1})$ , is

$$R = R_1 + R_4 + \frac{1}{\frac{1}{R_2} + \frac{2}{R_3}} \tag{5}$$

The heat flux, q (W), can be computed by using

$$q = (T_a - T_{LH2})/R \tag{6}$$

and the temperature of the inner and outer tank are

$$T_i = T_{LH2} + qR_4, \quad T_o = T_a - qR_1$$
 (7)

The heat flux transferring through  $R_2$  and  $R_3$ , denoted as  $q_2$  and  $q_3$ , respectively, are computed as

$$q_2 = (T_o - T_i)/R_2, \quad q_3 = (T_o - T_i)/R_3$$
 (8)

The boiling-off rate of  $LH_2$ , M (kg/s), is computed by using

$$M = q/h_{fq} \tag{9}$$

Note that the heat flux to LH<sub>2</sub> prior to phase change is ignored since the amount of heat is very small compared to the latent heat.

At a 65000 ft altitude, the ambient air temperature is  $-70^{\circ}$  F (216.7 K) and the pressure is 0.826 psia. The LH<sub>2</sub> is filled in at a temperature of 20.4 °K. At a 30 psia internal pressure, the boiling temperature of LH<sub>2</sub> is 23 °K and the latent heat,  $h_{fg}$ , is 428 kJ/kg. In Eq. (1),  $h_1$  is chosen as 1 W/m<sup>2</sup>-K by using the equation for free convection around a sphere [4], and  $\epsilon = 0.02$  for polished aluminum surface. In Eq. (2),  $k_e = 0.00016$  W/m-K for MLI and 0.0017 W/m-K for aerogel [4]. In Eq. (3),  $k_2 = 12.6$  W/m-K for stainless steel AISI 304 and 163 W/m-K for aluminum,  $L_2 = 2$  in. and  $A_{c,2} = 0.44\pi$  in.<sup>2</sup>, and  $k_1 = k_3 = k_4 = 163$  W/m-K for aluminum inner and outer tanks and support rod. In Eq. (4),  $h_4$  is chosen as 10 W/m<sup>2</sup>-K based on the table data for free convection of liquid [4].

In the first case, it is assumed that no heat leak through the ring by conduction, that is,  $R_3 = \infty$ . The heat flux and thermal resistance of the tank system is listed in Table 1. It can be seen that the thermal resistance between the hydrogen and inner tank and that between the ambient air and outer tank are very small. Replacing MLI with aerogel insulation, the thermal resistance,  $R_2$ , increases almost 10 times. The total heat flux is also increased almost 8 times. In the second case, it is assumed that the ring is made of stainless steel, the thermal resistance of the ring becomes dominant in  $R_3$  and the resistance of aluminum tanks and support is negligible, which gives

$$R_3 \approx \frac{1}{k_2 A_{c,2} / L_2}$$
 (10)

The calculated results of the thermal resistances and heat flux for both MLI and aerogel insulation are listed in Table 2. It can be seen that the total heat flux is increased to 97.8 W from 13.3 W for MLI and to 217 W from 111 W for aerogel insulation. It is obvious that the heat loss through the penetration by conduction is significant. The total heat flux increases 7.35 times and 86% heat loss through the rings for MLI and 2 times and 38% heat leaks through the rings for aerogel. For the tank system with aluminum rings, the 1D result is not available due to the thermal resistances of the inner/outer tanks and supports can not be estimated, which will be investigated in the three-dimensional thermal analysis as follows.

### III. Three-dimensional FEA of the Tank System

The tank system with MLI or aerogel insulation and aluminum or stainless steel spacer rings are analyzed using MSC Pthermal. A part of the finite element model of the tank system that has 73,494 nodes and 213,955 tetra elements is shown in Fig. 4. The free convection boundary condition is imposed on the inner surface of inner tank and the outer surface of the support inside the tank. The heat transfer coefficient is chosen as  $5 \text{ W/m}^2\text{-K}$  for the gaseous hydrogen part and  $10 \text{ W/m}^2\text{-K}$  for the liquid hydrogen part based on the table data for free convection in [4]. The part of the support outside the tank is assumed to be adiabatic. Free convection and radiation boundary conditions are imposed on the outer surface of the outer tank in the same way as that in one-dimensional analysis. On the outer surface of the inner tank, the effective heat transfer coefficient,  $h_e = k_e/t$  with t being the thickness of the insulation, was used, and  $h_e = 0.00314 \text{ W/m}^2\text{-K}$  for MLI and  $0.03346 \text{ W/m}^2\text{-K}$  for aerogel. There is no contact between the support and outer tank. In the case with the rings and support rod, the support and inner tank, the rings and inner/outer tanks are assumed to have no thermal resistance between each of them. Otherwise, larger thermal resistances are imposed at the

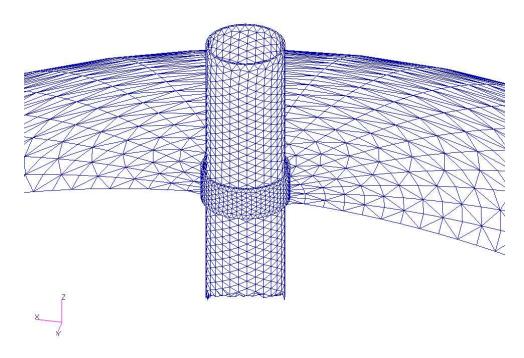


Figure 4. Part of the finite element model of the tank system.

interface to represent no heat transfer through there. Three cases including (i) without rings and support; (ii) with support and aluminum rings; and (iii) with support and stainless steel rings are considered here for both MLI and aerogel insulation.

To show the representative temperature distribution on the tank assembly, the MSC Pthermal results of the tank system with evacuated MLI blanket and aluminum spacer rings are plotted in Figs. 5 to 12. The temperature contour for the tank system is shown in Fig. 5, showing that most part of the outer tank is almost as hot as the ambient and most of the inner tank is nearly as cold as the hydrogen. A large temperature gradient appears around the top and bottom areas where the support penetrates. The close view of the top area is shown in Fig. 6. It can be seen that the heat conducts from the hot outer tank to the connecting ring and the inner tank then goes to the support. The hottest spot on the top part of the support is located at the interface of the support and inner tank, showing a temperature of 126 °K. The similar phenomena is seen at the bottom part plotted in Fig. 7. However, the bottom part has a lower temperature due to the contact with liquid hydrogen instead of gaseous hydrogen. The maximum temperature on the bottom part of the support is around 114 °K. The top ring is around 10 °K hotter than the bottom ring. The observation described above is further verified by the plots of the temperature contour on the top/bottom ring (together with part of the inner and outer tanks) in Figs. 8 and 9, and on the support (together with part of the inner tank) in Fig. 10. The temperature contour of the inner and outer tanks are plotted in Figs. 11 and 12, respectively, to show a clear view of the temperature distribution. The corresponding temperature contours for the tank system with aerogel insulation are plotted in Figs. 13 to 15. The temperature distribution is very similar to those presented above for the tank system with evacuated MLI. However, the outer tank is much hotter and the inner tank is much cooler because much less heat transfers from the outer tank to the inner tank. The temperature distribution of the tank system with stainless steel spacer rings is very similar to that with aluminum spacer rings and is not plotted here.

The three-dimensional results of the total heat flux are listed and compared with the one-dimensional

result in Table 3 for the three cases. For the system without a support and rings, the heat flux transferring into the LH<sub>2</sub> is 16.5 W for MLI and 137 W for aerogel, which agrees well with the one-dimensional results that are 13.3 W and 111 W, respectively. For the system with stainless steel rings, the total heat loss is 87.5 W for MLI and 213 W for aerogel, which is very close to the one-dimensional results that are 97.8 W and 217 W. For the system with aluminum rings, the total heat flux (237 W) for MLI is 2.7 times of that with stainless steel rings, and while for aerogel, the heat flux (354 W) becomes 1.7 times. It can be concluded that the system with stainless steel spacer rings and MLI blanket results in the least heat flux. The corresponding boiling-off rate of LH<sub>2</sub>, M, is 1.62 lbm/hr. For a 14-day mission, an extra 544 lbm mass of LH<sub>2</sub> has to be carried for boiling-off. However, the tank is designed to carry 1,300 lbm LH<sub>2</sub> for the required power. With the total boiling-off mass of the LH<sub>2</sub> within the 14-day mission being taken into an account, the vehicle can operate 8 to 9 days on 1,300 lbm LH<sub>2</sub> instead of 14 days. The heat leak from the tank system with stainless steel rings and MLI blanket needs to be further reduced to minimize the boiling-off rate of LH<sub>2</sub>. A 81.1% of total heat loss still goes through the penetration by conduction. The ring between the inner and outer tank could be replaced with a bellows made of stainless steel to increase its thermal resistance. The support rod could be modified to have two separate parts located at the top and bottom parts of the tank and ended at the inner tank such that the heat transfer area between the support and hydrogen is greatly reduced. Other solutions that might be pursued to minimize the heat flux transferring to the hydrogen from the ambient are not the focus of this paper and will not be further discussed here.

#### IV. Conclusion

The thermal analyses on the LH<sub>2</sub> tank system used in an unmanned aerial vehicle were performed. Both one-dimensional thermal model and three-dimensional FEA results are presented and discussed. The computed results of the heat flux transferring to the tank system showed that the heat leak through penetrations are significant. For a 14-day mission, the current tank design with MLI insulation requires an extra 544 lbm LH<sub>2</sub> for boiling-off while the tank carries only 1,300 lbm LH<sub>2</sub> for the required power. The vehicle can operate 8 to 9 days when taking into an account of the boiling-off of the LH<sub>2</sub>. The heat flux transferring to the tank system from the ambient has to be further reduced. The thermal resistance between the inner and outer tanks should be maximized by reducing the contact surface area between the ring and the inner/outer tank and by replacing the ring with a bellows made of stainless steel if it is possible.

#### References

<sup>&</sup>lt;sup>1</sup> Sullivan, R., et. al, Engineering Analysis Studies for Preliminary Design of Light Weight Cryogenic Hydrogen Tanks in UAV Applications, NASA TP-2006-214094, 2006.

<sup>&</sup>lt;sup>2</sup>Colozza, A., **Hydrogen Storage for Aircraft Applications Overview**, NASA/CR-2002-211867.

<sup>&</sup>lt;sup>3</sup>ESA-CR(P)-754, Vol. 2, **Spacecraft Thermal Control Design Data, Vol. 2**, Polytechnical University of Madrid (Spain).

<sup>&</sup>lt;sup>4</sup>Incropera, F.P. and DeWitt, D.P., **Fundamentals of Heat and Mass Transfer**, John Wiley and Sons, 1990.

Table 1. One-dimensional thermal results (without support and rings)

$\phantom{aaaaaaaaaaaaaaaaaaaaaaaaaaaaaaaaaaa$	$R_1$	$R_2$	$R_3$	$R_4$	q	$q_2$	$2q_3$
$(\mathrm{W/m}\text{-}\mathrm{K})$	$(W/K)^{-1}$	$(W/K)^{-1}$	$(W/K)^{-1}$	$(W/K)^{-1}$	(W)	(W)	(W)
$0.00016^{1}$	0.047	14.52	$\infty$	0.0047	13.3	13.3	0
$0.0017^2$	0.030	1.37	$\infty$	0.0047	111	111	0

<sup>&</sup>lt;sup>1</sup>MLI, <sup>2</sup>Aerogel

Table 2. One-dimensional thermal results (with support and stainless steel rings)

$\phantom{aaaaaaaaaaaaaaaaaaaaaaaaaaaaaaaaaaa$	$R_1$	$R_2$	$R_3$	$R_4$	q	$q_2$	$2q_3$
$(\mathrm{W/m}\text{-}\mathrm{K})$	$(W/K)^{-1}$	$(W/K)^{-1}$	$(W/K)^{-1}$	$(W/K)^{-1}$	(W)	(W)	(W)
0.00016	0.047	14.52	4.52	0.0047	97.8	13.2	84.6
0.0017	0.030	1.37	4.52	0.0047	217.0	135.1	81.9

Table 3. Three-dimensional thermal results compared with the one-dimensional results

	Without support		With support and		With support and	
	and rings		stainless steel rings		aluminum rings	
$k_e$	3D FEA   1D result		3D FEA	1D result	3D FEA	
(W/m-K)	q (W)	q(W)	q (W)	$q (\mathrm{W})$	$q~(\mathrm{W})$	
0.00016	16.5	13.3	87.5	97.8	237	
0.0017	137	111	213	217	354	

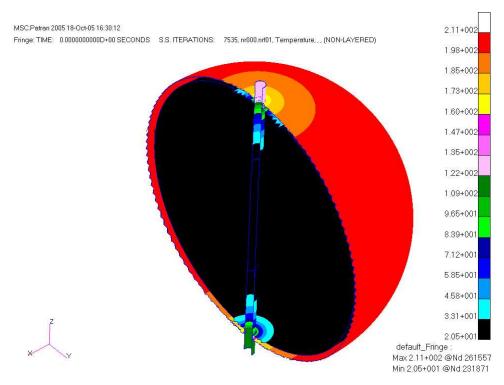


Figure 5. Temperature (K) contour on the tank system with MLI.

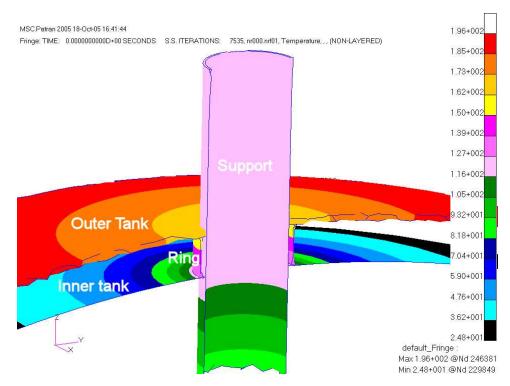


Figure 6. Close view of the top part of the tank system with MLI.

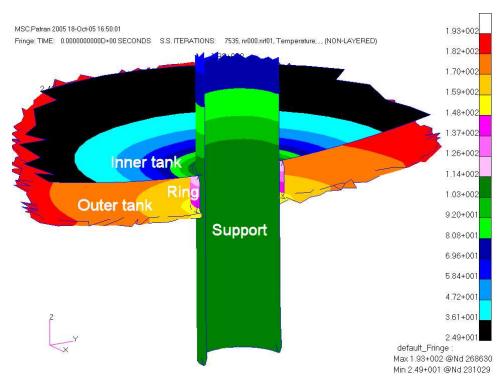


Figure 7. Close view of the bottom part of the tank system with MLI.

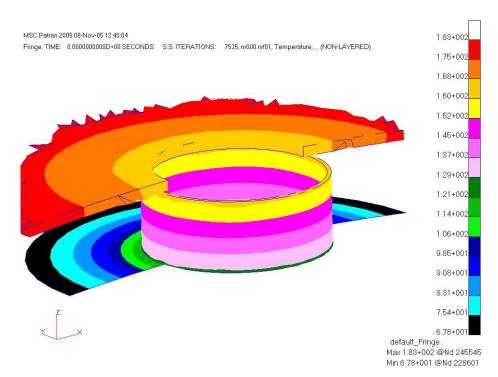


Figure 8. Temperature (K) contour of the top ring with part of the inner/outer tank (MLI).

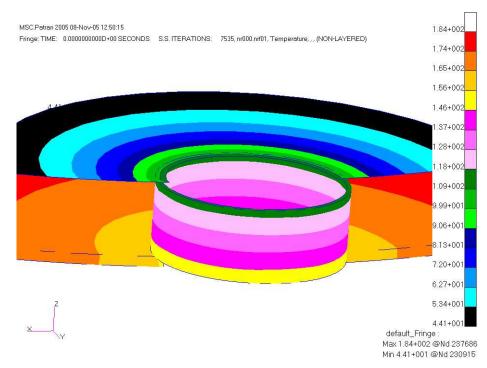


Figure 9. Temperature (K) contour of the bottom ring with part of the inner/outer tank (MLI).

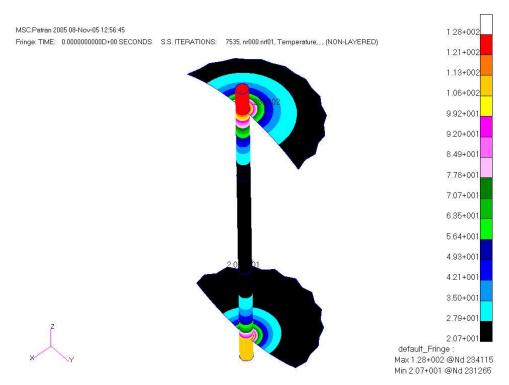


Figure 10. Temperature (K) contour of the support rod with part of the inner tank (MLI).

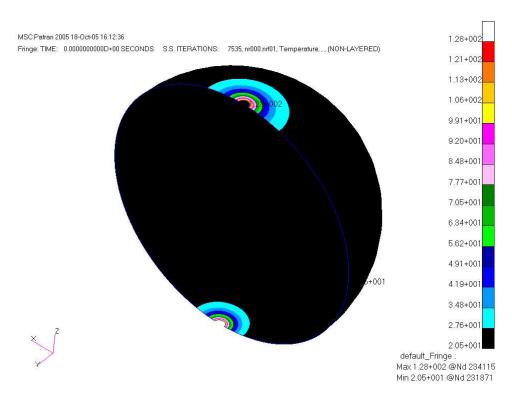


Figure 11. Temperature (K) contour of the inner tank (MLI).

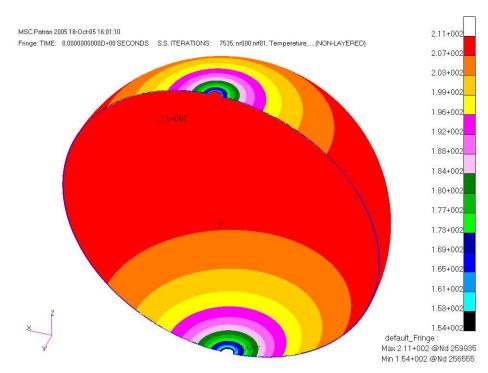


Figure 12. Temperature (K) contour of the outer tank (MLI).

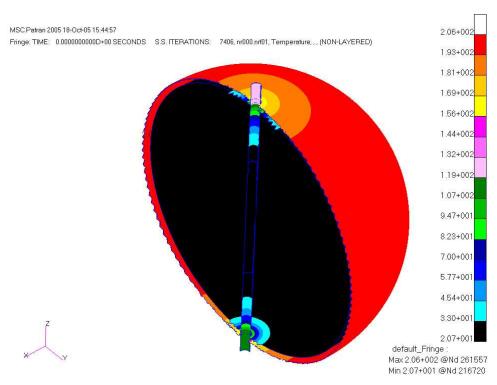


Figure 13. Temperature (K) contour on the tank system with aerogel insulation.

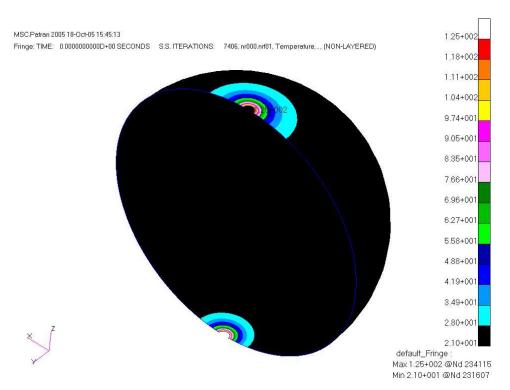


Figure 14. Temperature (K) contour of the inner tank (aerogel insulation).

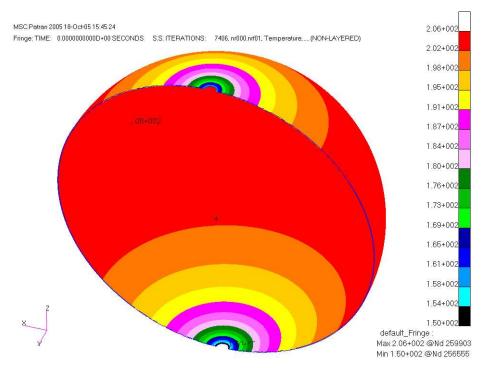


Figure 15. Temperature (K) contour of the outer tank (aerogel insulation).

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17.	SECURITY CLASSIFICATION	18. SECURITY CLASSIFICATION	19. SECURITY CLASSIFICATION	20.	LIMITATION OF ABSTRACT
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